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**Key indicators**

Single-crystal X-ray study  
 T = 298 K  
 Mean  $\sigma(C-C)$  = 0.005 Å  
 R factor = 0.069  
 wR factor = 0.204  
 Data-to-parameter ratio = 12.8

For details of how these key indicators were  
 automatically derived from the article, see  
<http://journals.iucr.org/e>.

# 1-[2,6-Dichloro-4-(trifluoromethyl)phenyl]- 5-[(2-furyl)methyleneamino]-1H-pyrazole- 3-carbonitrile

The title compound, C<sub>16</sub>H<sub>7</sub>Cl<sub>2</sub>F<sub>3</sub>N<sub>4</sub>O, is a tricyclic imide with an overall U-shape. There are  $\pi$ - $\pi$  interactions between the pyrazole and furyl rings.

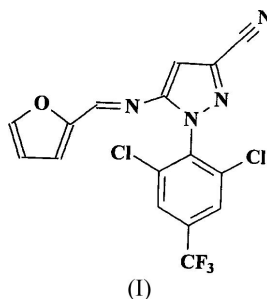
Received 28 February 2005

Accepted 11 August 2005

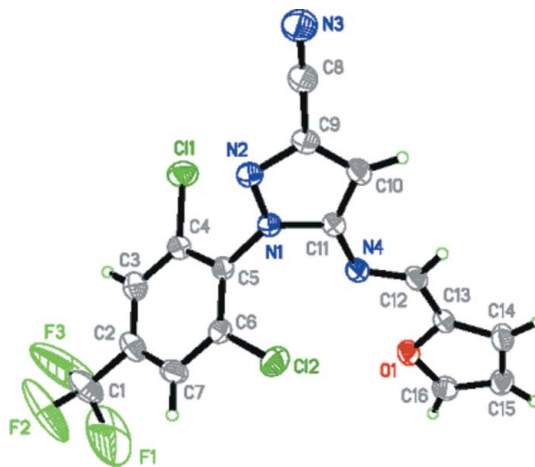
Online 17 August 2005

**Comment**

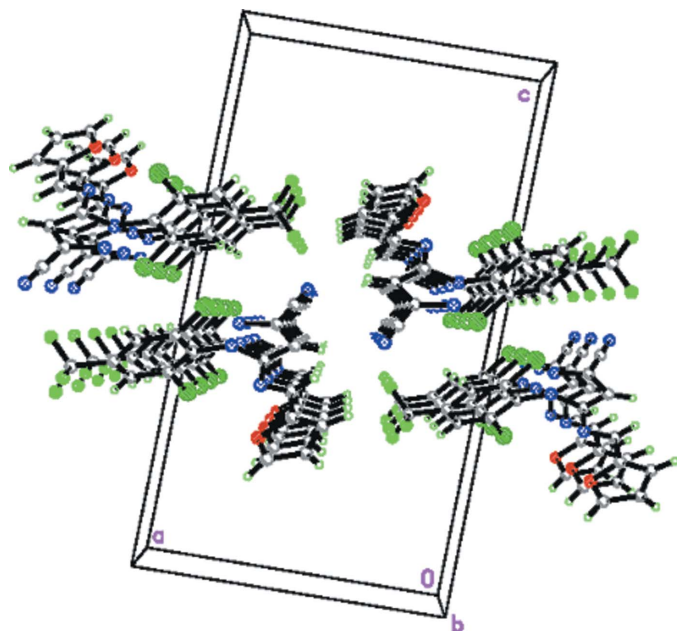
The title compound, (I), is an important starting material for the synthesis of 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-4-(trifluoromethyl)thiopyrazole, 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-4-(trifluoromethylsulfenyl)pyrazole and 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-4-(trifluoromethylsulfonyl)pyrazole, which are all good insecticides (Hatton *et al.*, 1993).



The structure of (I) is shown in Fig. 1, with the atom-numbering scheme. The molecule contains three planar moieties, forming an overall U-shape. The dihedral angles between the pyrazole and the furyl and benzene rings are



**Figure 1**  
 The structure of (I), showing the atomic numbering scheme and displacement ellipsoids drawn at the 50% probability level.



**Figure 2**  
The packing of (I), viewed down the *b* axis.

19.8 (2) and 67.9 (1)°, respectively. The plane-to-plane separation of 3.8411 (1) Å between the furyl and pyrazole rings indicates the presence of a weak  $\pi$ - $\pi$  interaction. In the crystal structure, the molecules are stacked along the *b* axis, as shown in Fig. 2.

## Experimental

Following the method of Hatton *et al.* (1993), reaction of 2,6-dichloro-4-(trifluoromethyl)amine with a suspension of nitrosyl sulfuric acid, followed by reaction with a solution of ethyl 2,3-dicyanopropionate in acetic acid, gave 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]pyrazole, which was then reacted with 2-furalan to give (I). Single crystals suitable for X-ray analysis were obtained by slow evaporation of a methanol solution (m.p. 449–451 K). IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3129, 2240, 1611, 1558, 1395, 1310, 1133, 873, 818;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  8.81 (*s*, 1H), 8.12 (*s*, 2H), 7.83 (*s*, 1H), 7.24 (*m*, 2H), 6.69 (*m*, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ): 154.2 (1C), 153.3 (1C), 152.1 (1C), 149.2 (1C), 136.6 (1C), 134.4 (*q*,  $J = 34.3$  Hz, 1C), 128.2 (1C), 127.05 (1C), 127.01 (1C), 126.95 (1C), 126.91 (1C), 123.3 (*q*,  $J = 271.6$  Hz, 1C), 122.0 (1C), 114.2 (1C), 114.0 (1C), 98.4 (1C).

### Crystal data

$\text{C}_{16}\text{H}_7\text{Cl}_2\text{F}_3\text{N}_4\text{O}$   
 $M_r = 399.16$   
Monoclinic,  $P2_1/n$   
 $a = 11.8828$  (9) Å  
 $b = 6.7072$  (5) Å  
 $c = 21.1191$  (16) Å  
 $\beta = 92.084$  (1)°  
 $V = 1682.1$  (2) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.576$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 3298 reflections  
 $\theta = 3.2$ – $25.0$ °  
 $\mu = 0.43$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
Block, colorless  
0.38 × 0.31 × 0.29 mm

### Data collection

Bruker APEX area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
Absorption correction: multi-scan (SADABS; Bruker, 2002)  
 $T_{\text{min}} = 0.854$ ,  $T_{\text{max}} = 0.885$   
8571 measured reflections

3013 independent reflections  
2571 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.019$   
 $\theta_{\text{max}} = 25.2$ °  
 $h = -13 \rightarrow 14$   
 $k = -8 \rightarrow 6$   
 $l = -24 \rightarrow 25$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.069$   
 $wR(F^2) = 0.204$   
 $S = 1.05$   
3013 reflections  
235 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1164P)^2 + 2.0094P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 1.12$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.64$  e Å<sup>-3</sup>

**Table 1**

Selected geometric parameters (Å, °).

C11–C4	1.724 (4)	N4–C12	1.273 (4)
F1–C1	1.271 (9)	N4–C11	1.382 (4)
O1–C16	1.352 (4)	C9–C10	1.389 (5)
O1–C13	1.362 (4)	C10–C11	1.374 (5)
N1–N2	1.348 (4)	C13–C14	1.350 (5)
N1–C11	1.374 (4)	C14–C15	1.403 (5)
N2–C9	1.337 (5)	C15–C16	1.340 (6)
N3–C8	1.147 (5)		
C16–O1–C13	106.2 (3)	C11–C10–C9	104.9 (3)
N2–N1–C11	113.0 (3)	C10–C11–N1	105.7 (3)
C9–N2–N1	103.2 (3)	C14–C13–O1	109.3 (3)
F3–C1–F2	111.0 (7)	C13–C14–C15	107.6 (3)
N3–C8–C9	179.1 (5)	C16–C15–C14	105.6 (3)
N2–C9–C10	113.1 (3)	C15–C16–O1	111.4 (3)

All H atoms were initially observed in a difference Fourier map and were then placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H = 0.93 Å and  $U_{\text{iso}}(\text{H}) = 1.22_{\text{eq}}(\text{C})$ . The low  $U_{\text{eq}}$  value of atom C1 compared with its neighbours may be attributed to the three possibly disordered F atoms. The highest peak is located 1.27 Å from atoms C1 and F.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Bruker, 2002); software used to prepare material for publication: *SHELXL97*.

This work was supported by the National Nature Science Foundation of China (No. 20272043) and the Nature Science Foundation of Zhejiang Province (No. M203001).

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